Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

Hai-Tao Zhang, Jing-Kang Wang,* Ying Chen and Lina Zhou

School of Chemical Engineering and Technology, Tianjin University, Tianjin 300072, People's Republic of China

Correspondence e-mail: cheny_79@yahoo.com.cn

Key indicators

Single-crystal X-ray study T = 293 K Mean σ (C–C) = 0.002 Å R factor = 0.046 wR factor = 0.127 Data-to-parameter ratio = 12.6

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e. The molecule of the title compound, $C_{10}H_{20}N_2O_4$, lies on an inversion centre. The crystal structure is stabilized by intermolecular N-H···O classical hydrogen bonds, which pack the molecules along the short [100] axis.

Dimethyl N,N'-(hexane-1,6-diyl)dicarbamate

Comment

The title molecule, (I), is an important carbamate compound, which serves as an intermediate in the preparation of *N*-aliphatic isocyanates (Wang *et al.*, 2006). It was first reported in 1967 for the synthesis of alkyl (6-isocyanatohexyl)-carbamates and dialkyl hexane-1,6-dicarbamates (Sinyavskii *et al.*, 1967). Since then, its ¹H NMR spectrum has been reported (Degner & Hannebaum, 1987) but, until now, its single-crystal structure has remained undetermined.

The molecular structure of (I) is shown in Fig. 1. It exhibits a classical aliphatic chain, similar to that found in normal paraffin hydrocarbons, with an inversion centre located at the mid-point of the C5–C5ⁱ bond [symmetry code: (i) 1 - x, -y, -z]. The C2–N1–C3 angle of 120.59 (12)° is suggestive of sp^2 hybridization for the N atom. Intermolecular hydrogen bonding involving the amine NH groups as donor and the carbonyl O atoms as acceptor (Table 1) link the molecules into infinite chains (Fig. 2) running along the short [100] axis.

Experimental

Hexane-1,6-diamine (10 ml) was reacted with dimethyl carbonate (75 ml) over an alkaline catalyst at atmospheric pressure. The reaction mixture was refluxed for 3 h. The excess of dimethyl carbonate was removed by evaporation under reduced pressure at 298 K. The crude product was purified by column chromatography and recrystallized from dimethyl carbonate. Colourless flake-like single crystals of (I) were obtained by slow evaporation of a dimethyl carbonate solution at 298 K (m.p. 385.2 K).

Z = 2

 $D_x = 1.235 \text{ Mg m}^{-3}$ Mo K α radiation $\mu = 0.10 \text{ mm}^{-1}$ T = 293 (2) KPlate, colourless $0.32 \times 0.22 \times 0.09 \text{ mm}$

| Crystal data |
|-------------------------------|
| $C_{10}H_{20}N_2O_4$ |
| $M_r = 232.28$ |
| Monoclinic, $P2_1/n$ |
| $a = 5.0290 (10) \text{\AA}$ |
| b = 23.674 (5) Å |
| c = 5.2550 (11) Å |
| $\beta = 93.56 \ (3)^{\circ}$ |
| $V = 624.4 (2) \text{ Å}^3$ |

© 2006 International Union of Crystallography All rights reserved

Received 20 October 2006 Accepted 15 November 2006

organic papers

Data collection

Rigaku R-AXIS RAPID IP diffractometer ω scans Absorption correction: multi-scan (*ABSCOR*; Higashi, 1995) $T_{min} = 0.970, T_{max} = 0.991$

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.046$ $wR(F^2) = 0.127$ S = 1.041432 reflections 114 parameters Only H-atom coordinates refined

Table 1

Hydrogen-bond geometry (Å, °).

| $D - H \cdots A$ | D-H | $H \cdot \cdot \cdot A$ | $D \cdots A$ | $D - H \cdots A$ |
|---------------------------|------------|-------------------------|--------------|------------------|
| $N1 - H1D \cdots O2^{ii}$ | 0.855 (19) | 2.116 (19) | 2.9664 (17) | 172.9 (17) |

6072 measured reflections 1432 independent reflections

 $R_{\rm int} = 0.070$

 $\theta_{\rm max} = 27.5^{\circ}$

1023 reflections with $I > 2\sigma(I)$

 $w = 1/[\sigma^2(F_0^2) + (0.0521P)^2]$

where $P = (F_0^2 + 2F_c^2)/3$

Extinction correction: *SHELXL97* Extinction coefficient: 0.055 (16)

+ 0.0675P]

 $(\Delta/\sigma)_{\rm max} = 0.001$

 $\Delta \rho_{\text{max}} = 0.13 \text{ e } \text{\AA}^{-3}$ $\Delta \rho_{\text{min}} = -0.16 \text{ e } \text{\AA}^{-3}$

Symmetry code: (ii) x - 1, y, z.

All H atoms were found in a difference map and refined with free coordinates. Isotropic displacement parameters for H atoms were fixed at $U_{\rm iso}({\rm H}) = 1.2U_{\rm eq}({\rm carrier atom})$ for methylene CH₂ and amine NH groups, and at $U_{\rm iso}({\rm H}) = 1.5U_{\rm eq}({\rm C})$ for the methyl group. Methyl H atoms were found to be disordered over two sites, due to rotation about the O1-C1 σ bond. Their site occupancy factors were constrained to sum to unity and converged to 0.53 (4) and 0.47 (4).

Data collection: *RAPID-AUTO* (Rigaku/MSC, 2004); cell refinement: *RAPID-AUTO*; data reduction: *CrystalStructure* (Rigaku/ MSC, 2004); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997*a*); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997*a*); molecular graphics: *SHELXTL* (Sheldrick, 1997*b*); software used to prepare material for publication: *SHELXL97*.

The authors gratefully acknowledge support from Tianjin University.

References

Degner, D. & Hannebaum, H. (1987). DE Patent No. 3606478 A1.

Higashi, T. (1995). ABSCOR. Rigaku Corporation, Tokyo, Japan.

Rigaku/MSC (2004). *RAPID-AUTO* and *CrystalStructure*. Rigaku/MSC Inc., The Woodlands, Texas, USA.



Figure 1

The molecular structure of (I), showing 30% displacement ellipsoids (arbitrary spheres for the H atoms). Both disorder components are shown. [Symmetry code: (A) 1 - x, -y, -z.]



Figure 2

The molecular packing of (I), viewed along [001]. Dashed lines indicate hydrogen bonds.

Sheldrick, G. M. (1997a). SHELXS97 and SHELXL97. University of Göttingen, Germany.

- Sheldrick, G. M. (1997b). SHELXTL. Bruker AXS Inc., Madison, Wisconsin, USA.
- Sinyavskii, V. G., Kravchenko, A. L., Lipatnikov, N. A. & Korsakova, L. N. (1967). Zh. Org. Khim. 3, 352–355. (In Russian.)
- Wang, G., Cheng, J. & Yao, J. (2006). CN Patent No. 1727330.